STRUCTURES OF THORNASTEROLS A AND B (BIOLOGICALLY ACTIVE GLYCOSIDES FROM ASTEROIDIA, XI)

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Summary: (20S)-thornasterol A, (20S,24R)-thornasterol B, (20S,24S)-thornasterol B, and (2OS)- 24-northornasterol A were stereoselectively prepared from (+)-asterone and were practically identical with the natural specimens. Consequently, the complete structures of versicosides B and C and acanthaglycoside F could be determined.

We have reported the isolation and structure of steroidal oligoglycoside sulfates from whole bodies of starfishes, <u>Asterias</u> amurensis [cf.] <u>versicolor</u> Sladen¹⁾ and <u>Acanthaster</u> planci L.²⁾. Versicosides B (1) and C (2)^{1b)} and acanthaglycoside F (3)^{2c)} contained thornasterol $B^{\prime\prime}$, 38,6 α ,20 ξ -trihydroxy-24 ξ -methyl-5 α -cholest-9(11)-en-23-one, as the steroidal component. However, they showed different negative Cotton maximums $([0]_{284}=-6028$ for 1, $\begin{bmatrix} \theta \end{bmatrix}_{287}$ =-6884 for 2, but $\begin{bmatrix} \theta \end{bmatrix}_{275}$ =-685 for 3) in the CD spectrum^{1b,2c} and also different chemical shifts corresponding to C-24 and C-28 of their aglycones in the 13 C-NMR spectrum. These results were attributed to the presence of (20S,24R)-thornasterol B and (2OS,24S) thornasterol B. We have now determined the configurations of the C-24 position in thornasterol B's and of C-20 in thornasterols A, B and 24-northornasterol A. The former was determined by preparing two thornasterol B's and the latter by leading 20E-hydroxy-23-oxocholest-5-en-36-yl p-bromobenzoate (13), which was prepared by the Mukaiyama's cross-aldol reaction, to 208hydroxycholesterol (15)⁴⁾.

Diacetyl (20S,24R)-thornasterols B (4) and (20S,24S)-thornasterol B (5) were synthesized as follows. (E)-4-Methyl-2-pentenol (7) was epoxidized by the Katsuki-Sharpless reaction⁵⁾,

- (5): R^3 =Me, R^4 = H diacetyl (20S)-thornasterol A
- $(6): R³ = H, R⁴ = H$

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using (+)-diethyl tartrate as a chiral element to an epoxy alcohol (<u>8</u>) (>90% e.e., ¹H-NMR of its MTPA ester) in a 64% yield. Treatment of 8 with trimethylaluminum in hexane at $0^{\circ}c^{7}$ afforded a 1,2-diol which was converted to an alcohol $(9)^{6}$ in a 48% yield from 8 by selective tosylation and subsequent LAH reduction. Swern oxidation of 9 to the corresponding ketone followed by enol etherification with trimethylchlorosilane gave (3R)-3,4-dimethyl-2-trimethylsilyloxy-1-pentene $(10)^{6}$ in a 56% yield from 9. 10 was treated with a mixture of diacetyl asterone^{1a,2a}) (11) and titanium tetrachloride in CH₂C1₂ at RT for 1 hr⁸) to give diacetyl (20S,24R)-thornasterol B (4)⁶⁾ in a 95% yield. The physical and spectroscopic properties of 4 were identical to those of the natural specimen¹) which was derived from versicoside B (1). Diacetyl (20S,24S)-thornasterol B (5) was prepared in a similar manner, except for the use of (-)-diisopropyl tartrate in the asymmetric epoxidation. The Mukaiyema's cross-aldol reaction of (3S)-3,4-dimethyl-2-trimethylsilyloxy-1-pentene (10')⁶⁾ with 11 gave 5^6 in excellent yield. The ¹³C-NMR chemical shifts of C-12 to C-28 as well as the ¹H-NMR shifts of 5 were superimposable on those of the aglycone moiety of acanthaglycoside F $\left(3\right) ^{2\text{C}}$. The CD spectra of 3 and 5 showed the same Cotton effect as shown in Figures 1 and 2.

Diacetyl (20S)-thornasterol A $(6)^{1a,2b,2c,6}$ and diacetyl (20S)-24-northornasterol A $(12)^{6,9}$ could be prepared from the corresponding trimethylsilyl enol ether and 11 by means of the Mukaiyama's cross-aldol reaction⁸⁾. 6 was identical to the natural specimen in every respect^{1a,2b)}. The 13 C-NMR chemical shifts of C-12 to C-27 of <u>12</u> and its characteristic 1 H-NMR peaks were in good agreement with those of the aglycone moiety of ophidianosides B and $\texttt{C}^\textsf{9}$.

In order to determine the C-20 configuration of thornasterols, 205-hydroxy-23-oxocholest-5-en-3β-yl p-bromobenzoate (13) $^{6)}$ and 205-hydroxy-23-oxo-24-norcholest-5-en-3β-yl acetate (14) ⁶⁾ were prepared from pregnenolone derivatives, in a similar manner. The conformation of the side chain, $C-20$ to $C-27$, of 13 and 14 were identical to that of 6 and 12, respectively, by comparison of the 13 C-NMR and 1 H-NMR chemical shifts of all the aldol products. In each case, the aldol reaction gave the single C-20 epimer having the 20-hydroxy-23-carbonyl moiety. These results could be well explained in terms of two factors, that the addition of organometallic reagents to the C-20 ketone of the steroid depends on the position and configuration of substituents near C-20 and the bulkiness of the reagent $^{10)}$. In the case of the reaction of pregnenolone acetate with a Grignard reagent, a 20s hydroxy compound is predominantly formed $^\mathcal{4)}$. On the base of the above results and physical constants of thornasterols and their analogs, 13 and 14 , the C-20 configuration was identical and apparently 20s. Conclusive evidence was obtained by chemical means. Sodium borohydride reduction of 13 led to a mixture of $C-23$ alcohols which were converted to acetates in a 71% yield from 13. The acetates were treated with lithium in EtNH₂ under reflux for 2 hr to give 208-hydroxycholesterol (15)⁶⁾ in a 36% yield and 208-hydroxycholest-5-ene (16)⁶⁾ in a 12% yield. <u>15</u> was acetylated to $17^{6)}$. The physical constants of <u>15</u> and <u>17</u> were in excellent agreement with the values in the literature $\overset{4)}{\cdots}$. Consequently, it was revealed that thornasterols had the same 20S configuration and the structures of versicosides B (1) and C (2) and acanthaglycoside F (3) were considered to have $(20S, 24R)$ -thornasterols B (4) and (20S,24S)-thornasterol B (5) as the aglycone component, respectively.

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References and Notes

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6) The following are physical constants of the products. <u>4</u>: colorless needles, m $(288, value)$, $$ mp 140-141°C, fD(c=O.O59,MeOH); [6125(nm)= 0(350), -5720 $-223(238)$, $-1250(216)$, $H-NMR(CDCl₂)$; $\delta = 0.78(s,13-Me)$, l.O2(s,10-Me), 1.33(s.20-Me), 2.58(s,22-2H), 0.93(d,J=7.OHz,26-or27-3H), 0.85(d,J=6.6Hz,26-or27-3H), 0.98(d,J=7.2Hz.24-Me). $5:$ colorless needles, mp 143-144°C, CD(c=0.077,MeOH); [θ|- (nm)= 0(338), +715(309), -631 $(277, \text{valley})$, $+69(242)$, $-215(226)$, $H-MMR(CDC1)$; $\delta = 0.78(s, 13-Me)$, $1.02(s, 10-Me)$, $1.34(s, 20-Me)$, $2.60(s, 22-2H)$, $0.92(d, J=6.6Hz, 28-0r27-3H)$, $0.85(d, J=6.8Hz, 26-0r27-3H)$, 0.99(d,J=5.5Hz,24-Me).
<u>6</u>: colorless needles, mp 153-154.5°C,_,CD(c=0.066,MeOH); [θ]²⁵(nm)= 0(340), -2450 $(288,\text{valley})$, +117(240), -1550(216), $\text{H-MMR}(\text{CDCl}_2)$; δ =0.77(s,13-Me), l.O2(s,10-Me), **1.33(s,20-Me),** 2.\$4(s,22-2H), 0.92(d,J=6.3Hz,26-and27-3H). $9:$ bp 135-140°C, "H-NMR(CDCl $_{3})$; ⁶=1.15(d,J=6.3Hz,3H), 0.91(d,J=7.2Hz,3H), $0.83(d,J=6.6\text{Hz},3\text{H}),$ $0.76(d,J=6.4\text{Hz},3\text{H}),$ $1.30(\text{m,1H}),$ $1.87(\text{m,1H}),$ $3.72(\text{m,1H}).$ $\underline{10}\colon$ bp 80°C/140mmHg, "H-NMR(CDCl $_2)$; $_2$ §=0.20(s,9H), C.84(d,J=6.6Hz,3H), 0.89(d,J=6.6Hz,3H), $3.97(s, 2H)$, $[\alpha]_D^{-\sim} = -11^{\circ}$ (c=2.0,EtOH). mp 158-159°C, CD(_{SE}O.O7O,MeOH); [θ]²⁵(nm)= 0(345), -2600 , $-1040(216)$, $[\alpha]_D^2 = +21^\circ(\text{c}=0.53,\text{CHCl}_2)$, IR(CCl₄); 3500, 1738, l.O2(s,lO-Me), l.&(s,20-Me?, 2.60(~,22-2H), 2.66(s,22-2H), $1.04(d,J=6.6Hz,26-0r27-3H)$, $1.06(d,J=7.0Hz,26-0r27-3H)$, H-NMR(CD OD); 6=0.80(s,13-Me), 1.04(s,lQJMe), 1.34(6,20-Me), , FD-MS; m/z= 502(M), 484(M – H₂O), 4 a1, 416(M - side chain), $\underline{13}\colon$ colorless needles, mp 167-169°C, [ɑ] $\stackrel{\sim}{\scriptscriptstyle{\cap}}$ 13-Me)**,** $= -12°(c=2.1, CHCl₂)$, $H-MMR(CDCl₂)$; $\delta = 0.87(s,$ 1.06(s,10-Me), 1.34(s,20-Me), 2.55(s,22-2H), 0.92(d,J=6.4Hz,26-and27-3H), anal. calcd. for C₃₄H₄₇O₄Br; C68.10, H7.90, found; C68.11, H7.93. $14:$ <code>H-NMR(CDCl $_2^{\frown})$; $\check{\;}$ 6=0.86(s,13-Me), 1.02(s,10-Me), 1.32(s,20-Me), 2.61(s,22-2H),</code> ~08(d,J=6.8Hz,~6-and21-3H). $15\colon$ mp $134.5-135.5$ °C, ^H-NMR(CDCl $_2)$; δ= $3.52(\text{m},3-1\text{H})$, 5.35 (d,J=4.8Hz,6-1H), 0.86(s,13 \cdot $1.01(\mathrm{s},10\text{--Me})$, $1\mathrm{_4}27(\mathrm{s},20\text{--Me})$, $0.87(\mathrm{d},\mathrm{J=6.3Hz},26\text{--} \mathrm{and}27\text{--}3\mathrm{H})$. $\underline{16}\colon$ mp <code>43-44°C, $\mathring{\;}$ H-NMR(CDC1 $_2)$; 6=5.27(d,J=4.4Hz,6-1H), O.86(s,13-Me), 1.00(s,10-Me),</code> 1.27(s,20-Me), 0.87(d,J₃g.3Hz,26-and27-3H), FD_TMS; m/z=386(M), 301(M - side chain). 17: mp 157.5-159°C, $\alpha \mid_{\mathsf{D}}^{\mathsf{D}} = -58$ °(c=0.97,CHCl₃), ⁺H-NMR(CDCl₃); 6=4.62(m,3-1H), 5.38 $(\texttt{d},\texttt{J=3.9Hz},\texttt{6-1H}), \ \texttt{0.86}(\texttt{\~{S}},\texttt{13-Me}), \ \texttt{1.02}(\texttt{s},\texttt{10-Me}), \ \texttt{1.27}(\texttt{s},\texttt{20-Me}), \ \texttt{0.87}(\texttt{d},\texttt{J=6.1Hz},\texttt{26-and})$ 27-3H). 7) W. R. Roush, M. A. Adam, and S. **M.** Peseckis, Tetrahedron Lett., 24, 1377 (1983). 8) T. Mukaiyama, K. Banno, and K. Narasaka, J. Am. Chem. Soc., 96 , 7503 (1974).

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